

Orthorhombic modification of triphenyltin 3-ureidopropionate

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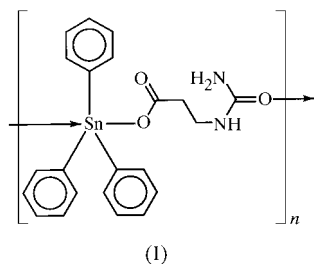
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The Sn atom in *catena*-poly[triphenyltin(IV)- μ -(3-ureidopropionato- $O^1:O^3$)], $[\text{Sn}(\text{C}_6\text{H}_5)_3(\text{C}_4\text{H}_7\text{N}_2\text{O}_3)]_n$, is five-coordinate and has a *trans*- C_3SnO_2 trigonal-bipyramidal geometry arising from bridging through the O atom of the ureido fragment of an adjacent carboxylate group. Infinite chains propagate helically along the *c* axis and adjacent chains are linked by $\text{N}-\text{H}\cdots\text{O}$ [$\text{N}\cdots\text{O}$ 2.851 (4) Å] hydrogen bonds into layers.

Comment

Triphenyltin alkanoates generally adopt carboxylate-bridged chain structures whose chains have either a flat or a helical conformation (Ng *et al.*, 1988; Tiekink, 1991, 1994). A strongly Lewis-basic site in the alkanoate group can compete successfully with the carboxyl O atom, and this is exemplified by the diethylphosphonoacetate (Ng & Kumar Das, 1994) and 3-diethylphosphonopropionate (Ng & Kumar Das, 1996). The



ureido O atom in the 3-ureidopropionate is probably more basic than the phosphoryl O atoms in 3-diethylphosphonopropionate [$\text{Sn}-\text{O}$ 2.116 (2) Å and $\text{Sn}\leftarrow\text{O}$ 2.397 (3) Å] as the dative bond [$\text{Sn}-\text{O}$ = 2.161 (2) Å and $\text{Sn}\leftarrow\text{O}$ 2.327 (2) Å] is much shorter than than in the 3-diethylphosphopropionate. Bond dimensions involving the Sn atom in the orthorhombic modification, (I), do not differ much from those found in the monoclinic modification (Lo *et al.*, 1991); the most significant difference is the almost linear $\text{O}-\text{Sn}\leftarrow\text{O}$ skeleton [$\text{O}-\text{Sn}\leftarrow\text{O}$ 179.2 (1)°], which contrasts with the bent skeleton [$\text{O}-\text{Sn}\leftarrow\text{O}$ 171.5 (1)°] of the monoclinic modification. The monoclinic ($P2_1/n$) modification which propagates in a zigzag

manner along the *c* axis, adopts a similar hydrogen-bonded [$\text{N}\cdots\text{O}$ 2.882 (4) Å] architecture.

Experimental

The crystal used for the X-ray measurements was a crystal that was selected from a batch of triphenyltin *N,N*-3-oxapentamethylene-thiocarbamoylthioacetate hydrate (Ng & Hook, 1999) crystals that had been kept in a bottle for several years. Whether the crystal had resulted from the solid-state decomposition of the hydrate or whether 3-ureidopropionic acid was a contaminant in the *N,N*-3-oxapentamethylene-thiocarbamoylthioacetic acid that reacted with triphenyltin hydroxide to form the 3-ureidopropionate could not be ascertained.

Crystal data

$[\text{Sn}(\text{C}_6\text{H}_5)_3(\text{C}_4\text{H}_7\text{N}_2\text{O}_3)]$
 $M_r = 481.11$
 Orthorhombic, $P2_12_12_1$
 $a = 11.2608$ (6) Å
 $b = 13.0527$ (5) Å
 $c = 14.6607$ (6) Å
 $V = 2154.9$ (2) Å³
 $Z = 4$
 $D_x = 1.483$ Mg m⁻³

Mo $K\alpha$ radiation
 Cell parameters from 25 reflections
 $\theta = 14.5$ – 15.0°
 $\mu = 1.209$ mm⁻¹
 $T = 298$ (2) K
 Cube, colorless
 $0.43 \times 0.43 \times 0.43$ mm

Data collection

Enraf–Nonius CAD-4 diffractometer
 ω scans
 Absorption correction: semi-empirical ψ scans [North *et al.* (1968) in *WinGX* (Farrugia, 1999)]
 $T_{\min} = 0.6661$, $T_{\max} = 0.7239$
 7014 measured reflections
 6262 independent reflections

5401 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.015$
 $\theta_{\text{max}} = 29.97^\circ$
 $h = -15 \rightarrow 15$
 $k = 0 \rightarrow 18$
 $l = 0 \rightarrow 20$
 3 standard reflections
 frequency: 60 min
 intensity decay: 3%

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.034$
 $wR(F^2) = 0.077$
 $S = 1.056$
 6262 reflections
 253 parameters
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.0412P)^2]$
 where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\text{max}} = 0.001$
 $\Delta\rho_{\text{max}} = 0.828$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.314$ e Å⁻³
 Absolute structure: Flack & Schwarzenbach (1988)
 Flack parameter = -0.04 (2); 2752 Friedel pairs

The structure was refined using 6262 reflections (3510 unique and an additional 2752 Friedel pairs); the chirality of the crystal used in the measurements was established by the Flack parameter, which refined to -0.04 (2).

Data collection: *CAD-4/PC* (Kretschmar, 1994); cell refinement: *CELDIM* in *CAD-4 VAX/PC Fortran System* (Enraf–Nonius, 1988); data reduction: *XCAD4* (Harms, 1997); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); software used to prepare material for publication: *SHELXL97*.

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